FIITJEEFARIDABAD

RANK IMPROVEMENT TEST SERIES JEE - Main - 2020

RITS - 22 CODE: 128150

Time: 3 hours Maximum marks: 300

INSTRUCTIONS

Caution: Question Paper CODE as given above MUST be correctly marked in the answer OMR sheet before attempting the paper. Wrong CODE or no CODE will give wrong results.

A. General Instructions

- 1. Attempt ALL the questions. Answers have to be marked on the OMR sheets.
- 2. The Test Booklet consists of **75** questions. The maximum marks are **300**.
- 3. There are *three* parts in the question paper. **Part-I** consisting of **Chemistry**, **Part-II** consisting of **Physics &Part-III** consisting of **Mathematics**. Each question is allotted **4** (**four**) marks for correct response.
- 4. Marking Scheme for All Two Parts:
 - (i) Part-A (01-20) Contains Twenty (20) multiple choice objective questions which have four (4) options each and only one correct option. Each question carries +4 marks which will be awarded for every correct answer and -1 mark will be deducted for every incorrect answer.
 - (i) Part-C (01-05) contains Five (05) Numerical based questions with single digit integer as answer, ranging from 0 to 9 (both inclusive). Each question carries +4 marks which will be awarded for every correct answer and there will be no negative marking.

Name of the Candidate:	
Batch :	Date of Examination
Enrolment Number :	

1. $CICH_2CHCI_2 \xrightarrow{alcoholic} X(major product)$

What will be the major product X of the above reaction?

(A) CHCI = CHCI

(B) $CH_2 = CCI_2$

(C) HC ≡ CH

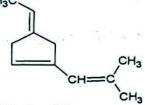
(D) CICH2CHO

2.

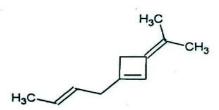
$$(Y) \xrightarrow{(1) O_3} (2) 2n/H_2O + H$$
Hydrocarbon $(2) 2n/H_2O$

Hydrocarbon (Y) in the above reaction can be?

(A) H



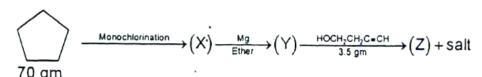
(B)



(C) H₃(

(D) All of these

3.



The number of moles of (Z) produced would be (assuming the yield to be 100%)?

(A) 1

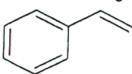
(B) 0.01

(C) 0.1

(D) 10 mol

Which of the following does not decolourise Br₂ water?

(A)



(B)



(C) HC = CH

(D)

- 5. Monomer used to prepare orion is:
 - (A) $H_2C = CH CI$
 - (C) H2C=CHCN

- (B) $F_2C = CF_2$
- (D) $H_2^2C = C CH = CH_2$

6.

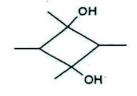
The major product formed in the above reaction is:

(A)

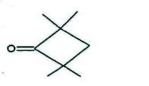


(B)

(C)



(D)



7. $CH_2 = C = CH_2 \xrightarrow{H_3O^+}$?

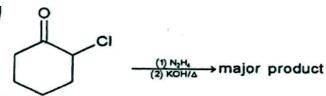
Select the correct statement(s) regarding the (major) product formed in the above reaction.

- (A) It reduces Fehling's solution
- (B) It is oxidised by Tollen's reagent

(C) It decolourises Br2 - CCl4

(D) It gives iodoform test

8.

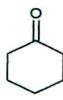


The major product formed in the given reaction is:

(A)



(B)



(C)



(D)



9. Which of the following method(s) does work for the preparation of secondary amine?

(A)
$$R_3N \xrightarrow{(i) Br-CN}$$
?

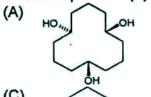
(C) RNC
$$\xrightarrow{\text{NI/H}_2}$$
?

(D) All of these



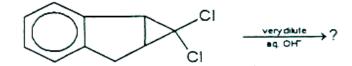
$$\xrightarrow{B_2H_8-THF} (X) \xrightarrow{H_2O_2+OH^-} (Y)$$

The final product (Y) is



(B)

11.



The major product is

(A)

(C)



(B)

(D) None of these

12.

The reagent used in the above transformation is:

- (A) (CH₃)₂SO
- (C) Acetone

- (B) $H C N(CH_3)_2$
- (D) P₄O₁₀

The major product formed is:

(A)

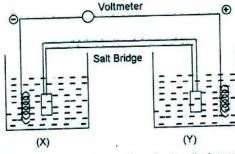
(B)

None of these (D)

Given-

The cell potential for the unbalanced chemical reaction:

 $Hg_2^{2+} + NO_3 + H_3O^+ \longrightarrow Hg^{2+} + HNO_2 + H_2O + e^-$



Is measured under standard conditions in the electrochemical sell shown in the diagram were 0.02 V.

- 14. Choose the correct statement for the given cell diagram
 - (A) Compartment (X) has less pH than compartment (Y)
 - (B) (Y) compartment has acidic solution
 - (C) Current will flow from X → Y through internal supply
 - (D) ΔG° for the above cell reaction is more than 1 at the equilibrium
- 15. Equilibrium constant for cell reaction is:

(A)
$$10^{3/2}$$

(B)
$$e^{-\frac{3}{2}}$$

(C)
$$e^{-\frac{2}{3}}$$

- 16. If same amount of charge (which is required for formation of 0.1 mol of HNO2 in the above given cell) used for electrolysis 0.1 M, 1 L aqueous solution of CuSO₄. Then the volume of gases liberated at STP will be:
 - (A) 2.24 litre

(B) 5.6 litre

(C) 22.4 litre

(D) 11.2 litre

Select the correct statements:

- (A) (Y) is
- (B) (Z) has three carbonyl groups
- (C) (Z) is OHC CHO
- (D) Formation of (X) involves equilibrium mixture of the iminium salt and the acylated enamine

Space for rough work

18. Select the correct order with respect to the mentioned properties:
(A) CH₃CH₂OH > (CH₃)₂ CHOH > (CH₃)₂ CDOH > (CH₃)₃ C.OH

(Ease of oxidation)

(D) PhSO₃H > PhCOOH > PhOH > PhCH₂ NH₃ (acidic strength)

Select the correct statement(s) regarding X, Y and Z:

20.) * Me
PhCONH₂ + H D
$$\xrightarrow{Bc_3 \cdot KOH}$$
 ?
CONH₂

Which of the following statement(s) is/are correct regarding the given reaction:

- (B) Me

 Cross over products PhNH₂ and H——D are also produced

 *NH₂
- (C) Migration of -R (alkyl group) of R CONH2 takes place with retention of configuration.
- (D) Me
 PhCONHBr and H—D are the intermediates formed
 CONHBr

- Select the total number of methods to convert cyclohexanone into 1,2-cyclohexanedione.
 - (I) nitrous acid and (aqueous medium)
 - (II) Br₂ + H₃O⁺; aqueous KOH; oxidation by CrO₃ in acetic acid
 - (III) aqueous NaOH + Δ; ozonolysis
 - (IV) SeO₂
 - (V) NaCN + dilute H₂SO₄; H₃O⁺; Δ
 - (VI) Br₂ + KOH (aqueous); Δ
 - (VII) Br₂ + H₃O⁺; N₂H₄; KOH + Δ; cold alkaline KMnO₄; HIO₄.
- 2. How many of the following compounds would undergo aldol condensation when treated with dilute aqueous caustic soda under suitable condition?

3. Find the total number of compounds which give yellow precipitate of iodoform when heated with l₂ and alkali?

- 4. What is the maximum number of moles of CH₃MgCl that can be consumed by one mole of phosgene?
- 5. How many geometrical isomers (excluding enantiomers) are possible for the compound given below?

Se	ction – 2			PHYSI
1.	straight lines. The distance from the	e boat A moves along the ri e buoy, the boats returned nd B respectively, if the velo	ver and the boat B across . Find the ratio t_{A}/t_{B} , wh	a river along mutually perpendicular the river. Having moved off an equal here t_A and t_B are the time of motion spect to water is 2 times greater than $ (D) \ 1/\sqrt{3} $
2. A massive platform of mass M is moving horizontally with speed $v = 6$ m/s. at $t = 0$, a body (m << M) is gently placed on the platform. If coefficient of friction between the body and platform is $\mu = 0.3$ and $g = 10$ m/s ² , then (A) The body covers a distance of 3 m on the platform in the direction opposite to the mot			g = 10 m/s², then	

- (B) The body covers a distance of 3 m on the platform in the direction of motion of the platform before
- (C) The body covers a distance of 6 m on the platform in the direction opposite to themotion of platform.
- (D) The body covers a distance of 6 m on the platform in the direction of motion of the platform before coming to rest.
- 3. A block is suspended vertically by an ideal spring of force constant K. If the block is pulled down by applying a constant force F and if maximum displacement of the block from its initial position of rest is δ ,

- A ball strikes a wall with a velocity \vec{u} at an angle θ with the normal to the wall surface and rebounds from it 4. at an angle β with the surface. Then:
 - (A) $(\theta + \beta) > 90^{\circ}$, if the wall is smooth(B) if the wall is smooth, coefficient of restitution = $\frac{\tan \beta}{\cot \theta}$
 - (C) if wall is smooth, coefficient of restitution $< \frac{\tan \beta}{\cot \theta}$ (D) none of the above

6.

- 5. A uniform rod AB of mass m and length ℓ is at rest on a smooth horizontal surface. A horizontal impulse P is applied to the end B perpendicular to the rod. The time taken by the rod to turn through a right angle is (A) $\frac{12\pi m\ell}{p}$ (C) $\frac{5\pi m\ell}{12P}$
 - (B) $\frac{\pi m \ell}{6P}$ A cubical box of side L is filled with two liquids of densities p and 2p. Each liquid fills half the volume of the
 - vessel. Then the net force on any one vertical side wall of the vessel due to liquid is. [Neglect atm pressure] (A) $\frac{5}{8} \rho g L^3$ (C) $\frac{1}{2} \rho g L^3$ (D) $\frac{1}{9} \rho g L^3$ (B) $\rho g L^3$
- 7. A cylindrical vessel is filled with water upto height h. The vessel empties in time t sec. if the water is now filled upto height nh. the vessel will be emptied in time

med apto height i	in, the resser will be empth	ca iii tiiiite	
(A) ηt sec	(B) η ² t sec	(C) $\sqrt{\eta}$ t sec	(D) t/η sec

9.	Two tuning forks A and B sounded together give 8 beats per second. With an air resonance tube closed a one end, the two forks give resonance when the two air columns are 32 cm and 33 cm respectively. What will be the frequencies of the two forks?				
	(A) 264 Hz, 256 Hz	(B) 340 Hz, 358 Hz	(C) 222 Hz, 300 Hz	(D) 176 Hz, 184 Hz	
10.	vessels are P_A , T_A , V and	P_B , T_B , V respectively. The ν	vessels are now connected	ture and the volume in the d through a small tube. If P nich of the following relation	
	$(A)\frac{P}{T} = \left[\frac{P_A}{T_A} + \frac{P_B}{T_B}\right]$	(B) $\frac{P}{T} = \frac{1}{2} \left[\frac{P_A}{T_A} + \frac{P_B}{T_B} \right]$	(C) $\frac{P}{T} = 2 \left[\frac{P_A}{T_A} + \frac{P_B}{T_B} \right]$	(D) $\frac{P}{T} = \frac{3}{2} \left[\frac{P_A}{T_A} + \frac{P_B}{T_B} \right]$	
11.	embedded in a thermally	-	heat capacity s and mass	mall coil of resistance wire m. if the temperature of the	
	(A) $\frac{\text{ms }\Delta T}{C}$	(B) $\sqrt{2 \frac{\text{ms } \Delta T}{C}}$	(C) $\sqrt{2 \frac{\text{mC } \Delta T}{\text{s}}}$	(D) $\frac{\text{mC}\Delta T}{\text{s}}$	
12.	A particle of mass 1×1 the	0^{-26} kg and charge $+1.6 imes$	$3 \cdot 10^{-19}$ C travelling with $3 \cdot 10^{-19}$	a velocity 1.28×10^6 m/s in	
	are present such that $\boldsymbol{E}_{\boldsymbol{x}}$	$= E_y = 0$, $E_z = -102.4$ kVi	${\sf m}^{\text{-}1}$ and ${\sf B}_{\sf x}={\sf B}_{\sf z}=0$, ${\sf B}_{\sf y}=$	magnetic field of induction B 8×10^{-2} Wb m ⁻² . f the particle (in meters) at	
	(A) $(0, 0, 6.4)$	(B) (0, 6.4, 0)	(C) (0, 0, 3.2)	(D) (6.4, 0, 0)	
13.				pectively. Each wire carries a points, where the magnetic	
	$(A) y^2 = x$	$(B) x = y^2$	(C) x - y = 0	(D) x + y = 0	
14.	An ac source of angular frequency ω is fed across a resistor R and a capacitor C in series. The current registered is I. If now the frequency of source is changed to $\omega/3$ (but maintaining the same voltage), the current in the circuit is found to be halved. The ratio of reactance to resistance at the original frequency ω will be				
	(A) $\sqrt{\frac{3}{5}}$	(B) $\sqrt{\frac{5}{3}}$	(C) $\frac{3}{5}$	(D) $\frac{5}{3}$	
15.	A person uses +1.5 D gla	asses to have normal vision bject. What will be the ma		uses a 20 D lens as a simple , is he uses the microscope	
	(A) 6	(B) 3	(C) 10	(D) 9	
	Space for rough work				

Two rods of identical dimensions, with Young's Moduli Y_1 and Y_2 are used to form a composite rod. The equivalent Young's Modulus for the composite rod is (if the composite rod is also having the same

(C) $\frac{1}{2}$ (Y₁ + Y₂) (D) Y₁ + Y₂

8.

dimension as one of the original rod)

(B) $\frac{1}{2} \frac{Y_1 Y_2}{Y_1 + Y_2}$

16.	wavelength of the photo		mes λ_2 . The atom has De	om as a result of which the Broglie's wavelength λ_3 and
	(A) $\lambda_3 = \sqrt{\lambda_1 \lambda_2}$	(B) $\lambda_1 = \frac{\lambda_2 \lambda_3}{\lambda_2 + \lambda_3}$	(C) $\lambda_1 = \sqrt{{\lambda_1}^2 + {\lambda_3}^2}$	(D) $\lambda_3 = \sqrt{{\lambda_1}^2 + {\lambda_2}^2}$
17.		_		ary positron (hypothetically), positron), c=speed of light) (D) none of these
18.	A radioactive sample ha	as decay constant λ. The ra	te of production of nuclei	in the given sample is $\frac{9\lambda N_0^2}{N}$,

18. A radioactive sample has decay constant λ . The rate of production of nuclei in the given sample is $\frac{9\lambda N_0^2}{N}$, where N_0 is the number of radioactive nuclei in the sample at t = 0 and N is the number of radioactive nuclei in the sample at time t = t sec. Then the number of nuclei present in the radioactive sample at t $\rightarrow \infty$,

(given $N_0=10^6$ nuclei)

(A)
$$4 \times 10^6$$

(B)
$$5 \times 10^6$$

(C)
$$1 \times 10^6$$

(D)
$$3 \times 10^6$$

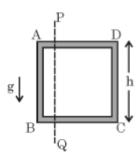
19. Consider the following data:

10 main scale division = 1 cm, 10 vernier division = 9 main scale division

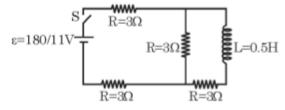
Zero of the vernier scale is right of the zero marking of the main scale with 6 vernier divisions coinciding with the main scale divisions and the actual reading for length measurement is 4.3 cm with 2 vernier divisions coinciding with main scale graduations. Then the actual length is

- (A) 4.26 cm
- (B) 4.32 cm
- (C) 4.38 cm
- (D) 4.60 cm
- 20. Which of the following statement regarding semiconductors is correct?
 - (A) An n type semiconductor is a negatively charged semiconductor
 - (B) Photodiodes are preferably used in reverse bias mode, as small changes in electron and hole concentration are better detected in reverse bias mode
 - (C) The voltage current variation graph for a diode has same units along positive and negative direction of the axis depicting current
 - (D) Threshold voltage for a germanium based diode is more than that of silicon based diode

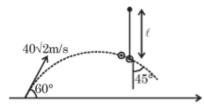
- Suppose a hypothetical magnetic field exists in space $\hat{B} = B_0 \hat{u}_r$ above the earth surface where \hat{u}_r is a unit vector directed radially outward from origin. Origin is on surface of earth. A light charged particle has to perform uniform circular motion in the combined uniform (vertical) gravitational field of earth and magnetic field with speed v and radius
 - r. Height of the plane of motion from earth surface will be $h=\frac{n\nu^2}{g}$. Find n.
- 2. ABCD is a square frame of conductor of electrical resistivity ρ. The frame lies in a vertical plane. PQ is an imaginary boundary separating space into two parts. Left of PQ, a uniform gravitational field g exists (figure) whereas no gravitational field is present right of PQ. The electrical potential difference between A and B will be k mgh/e e is charge on an electron and m is mass of electron. Find k. (Given the square frame is fixed).

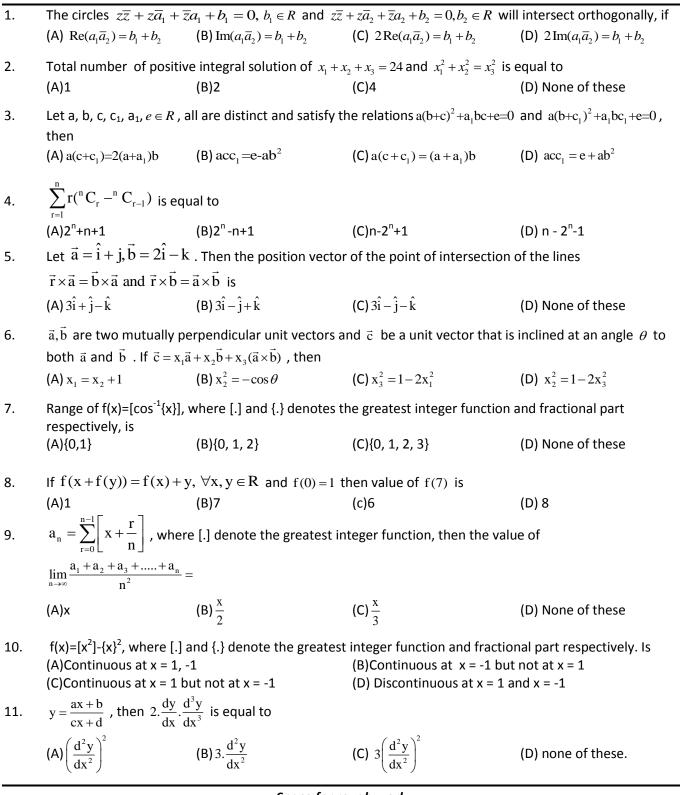


- 3. A shell of radius 1m is coated with a thin layer of β^- active material. It's initial charge is zero and initial number of active atoms is $\frac{4}{3} \times 10^{12}$. If half life of decay is 1 hr and all the electron are emitted with an energy of 1.44 keV, find the time (in hr.) after which charge on the sphere becomes constant. Neglect the time taken by electron to return back.
- In the circuit shown, find the current in Amp through the battery at t = 0.1 ln 2 sec, if switch is closed at t = 0.



A ball is projected with initial velocity 40√2 m/s at an angle of 60° with horizontal as shown in figure. When its velocity is making an angle of 45° with downward vertical, it collides a vertically hanging ball of same mass and sticks to it as shown in figure. Find the maximum value of length ℓ (in m) of the string, so that the combined mass completes the vertical circle in subsequent motion.





	(A) $\frac{a^2 - b^2}{2}$	(B) $2(a^2 - b^2)$	(C) $\frac{a^2 + b^2}{2}$	(D) $2(a^2+b^2)$
14.	$\int_{-\ell n3}^{\ell n3} \left \frac{e^x - 1}{e^x + 1} \right dx \text{ is equal t}$:0	2	
	(A) $\ell n \frac{4}{3}$	(B) $\ell n \frac{2}{3}$	(C) $2.\ell n \frac{4}{3}$	(D) $2.\ell n \frac{2}{3}$
15.	If $\cos A = \frac{3}{4}$, then the	value of expression 32.sin	$\frac{A}{2}$.sin $\frac{5A}{2}$ is equal to	
	(A)11	(B)-11	(C) $\sqrt{11}$	(D) −√11
16.	The exhaustive set of v	values of 'a' such that $x^2 + a$	$x + \sin^{-1}(x^2 - 4x + 5) + \cos^{-1}(x^2 - 4x + 5)$	5) = 0 has atleast one solution
	$(A) \left\{-2 - \frac{\pi}{4}\right\}$	$(B)\left\{-\infty,-2-\frac{\pi}{4}\right\}$	(C) $\left(-\infty, -2 - \frac{\pi}{4}\right]$	(D) $\left[-2-\frac{\pi}{4},-\infty\right)$
17.	In acute angled triangle	e ABC, $r = r_2 + r_3 - r_1$ and \angle	$B > \frac{\pi}{3}$, then exhaustive range	ge of $\frac{a-c}{b}$ is
	(r_1, r_2, r_3, r) and s have us			
	$(A)\left(\frac{1}{2},1\right)$	$(B)\left(\frac{1}{4},\frac{1}{2}\right)$	(C) $\left(\frac{1}{4},1\right)$	(D) $\left(\frac{1}{3},1\right)$
18.		n of straight lines that can	be obtained by reflecting the	e lines $y = x-2 $ in the y-
	axis, is (a) $y^2 + x^2 + 4x + 4 = 0$	(B) $y^2 + x^2 - 4x + 4 = 0$	(C) $y^2 - x^2 + 4x - 4 = 0$	(D) $y^2 - x^2 - 4x - 4 = 0$
19. Tangent drawn to $x^2 + y^2 = 25$ at the point P(3,4) meets the circle $x^2 + y^2 = 81$ at the points A and B, tangents drawn to $x^2 + y^2 = 81$ at the points A and B intersect at C, then co-ordinates of 'C' is.				
	,	(b) $\left(\frac{243}{25}, \frac{324}{25}\right)$, ,	(D) None of these
20. If the circle $x^2 + y^2 + 2ax + 2by + c = 0$ passes through exactly three quadrants and does not pass through				
	origin, then (A) $c > 0, a^2 > b^2$	(B) $c < 0, a^2 + b^2 > 2c$	(C) $c(a^2-c)(b^2-c) < 0$	(D) $c > 0, a^2 > c, b^2 > c$
		Space for I	rough work	
			-	

If the tangent to the curve xy+ax+by=0 at (1, 1) makes an angle $\tan^{-1}(2)$ with x-axis, Then $\frac{a+b}{ab}$ is equal to

(C) 1

(D) -1

(B) - 1/2

If a > b > 0, then maximum value of $\frac{ab(a^2-b^2)\sin x.\cos x}{a^2\sin^2 x+b^2\cos^2 x}$ in $x \in \left(0,\frac{\pi}{2}\right)$ is

12.

13.

(A) 1/2

PART-C

Integer Answer Type

This section contains 5 questions. The answer to each of the questions is a single-digit integer, ranging from 0 to 9.

- Let $f(x) = \tan^{-1} x \frac{2}{\pi} (\tan^{-1} x)^2 + \frac{4}{\pi^2} (\tan^{-1} x)^3 \dots$ upto infinite terms. If the equation $f^2(x) + (\sin^{-1} x)^2 = k$ posses a solution, then the number of integral values of k is
- If $f: R \to R$ satisfies the functional equation $f(x) + f\left(1 \frac{1}{x}\right) = \tan^{-1} x$, $\forall x \in R \{0\}$ and let $N = \int_0^1 f(x) dx$, then least integer greater than or equal to N is
- 3. Let y = f(x) be a thrice differentiable function defined on R such that f(x) = 0 has at least 5 distinct zeros, then minimum number of zeros of the equation f(x) + 6f'(x) + 12f''(x) + 8f'''(x) = 0 is
- 4. Let $N=2^{2015} imes \frac{\int\limits_0^1 x^{1007} (1-x)^{1007} \, dx}{\int\limits_0^1 x^{1007} \Big(1-x^{2016}\Big)^{1007} \, dx}$, then the number of divisors of N of the form 4n+2, $(n\in N)$ is
- A certain kind of bacteria either die, split into two or split into three bacteria. All splits are exact copies. The chances of dying is $\frac{1}{4}$, the chances of splitting into two is $\frac{1}{2}$ and splitting into three is $\frac{1}{4}$. If the probability that it survives for infinite length of time is $\frac{m-\sqrt{13}}{n}$ (m,n \in N), then the value of (m+n) is

RITS-22 JEE MAINS-2020 ANSWER KEY

Code:128150

CHEMISTRY		PHYSICS		MATHEMATICS	
1	В	1	Α	1	С
2	D	2	D	2	В
3	С	3	В	3	D
4	D	4	В	4	С
5	С	5	D	5	Α
6	А	6	Α	6	С
7	D	7	С	7	Α
8	С	8	Α	8	Α
9	D	9	Α	9	В
10	В	10	В	10	D
11	С	11	В	11	С
12	А	12	D	12	Α
13	В	13	С	13	Α
14	В	14	Α	14	С
15	D	15	D	15	Α
16	А	16	В	16	Α
17	С	17	Α	17	D
18	С	18	D	18	D
19	В	19	Α	19	В
20	А	20	В	20	D
1	3	1	1	1	5
2	5	2	3	2	2
3	7	3	2	3	2
4	3	4	2	4	5
5	8	5	4	5	7